according to 29CFR1910/1200 and GHS Rev. 3

**Initial preparation date: : 01.21.2015** 

#### **Cupric Chloride, Dihydrate**

# SECTION 1: Identification of the substance/mixture and of the supplier

**Product name**: Cupric Chloride, Dihydrate

Manufacturer/Supplier Article number: S25279A

Recommended uses of the product and restrictions on use: Laboratory

#### **Manufacturer Details:**

AquaPhoenix Scientific 860 Gitts Run Road, Hanover, PA 17331 (717) 632-1291

# **Supplier Details:**

Fisher Science Education 6771 Silver Crest Road, Nazareth, PA 18064 800 955-1177

# **Emergency telephone number:**

Emergency Telephone No.: 800-255-3924

#### **SECTION 2: Hazards identification**

#### Classification of the substance or mixture:



#### Irritant

Acute toxicity (oral, dermal, inhalation), category 4 Skin irritation, category 2 Eye irritation, category 2A



#### **Environmentally Damaging**

Acute hazards to the aquatic environment, category 1 Chronic hazards to the aquatic environment, category 1

Acute Tox. 4.

Skin Irrit. 2.

Eye Irrit. 2.

Aq AcTox 1.

Ag ChTox 1.

Hazards Not Otherwise Classified - Combustible Dust.

Signal word: Warning

#### **Hazard statements:**

Harmful if swallowed.

Causes skin irritation.

Causes serious eye irritation.

Very toxic to aquatic life.

Very toxic to aquatic life with long lasting effects.

# **Precautionary statements:**

If medical advice is needed have product container or label at hand.

Keep out of reach of children.

Read label before use.

Wash skin thoroughly after handling.

Avoid release to the environment.

according to 29CFR1910/1200 and GHS Rev. 3

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## **Cupric Chloride, Dihydrate**

Wear protective gloves/protective clothing/eye protection/face protection.

Use personal protective equipment as required.

Do not eat, drink or smoke when using this product.

Specific treatment (see supplemental first aid instructions on this label).

Rinse mouth.

Take off contaminated clothing and wash before reuse.

Collect spillage.

IF SWALLOWED: Call a POISON CENTER or doctor/physician if you feel unwell.

IF ON SKIN: Wash with soap and water.

If skin irritation occurs: Get medical advice/attention.

If eye irritation persists get medical advice/attention.

IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses if present and easy to do.

Continue rinsing. Store in a dry place.

Dispose of contents and container to an approved waste disposal plant.

Other Non-GHS Classification: None

## **SECTION 3: Composition/information on ingredients**

#### Ingredients:

Ingredients:				
CAS 10125-13-0	Cupric Chloride, Dihydrate	100 %		
	Po	ercentages are by weight		

#### **SECTION 4: First aid measures**

## **Description of first aid measures**

#### After inhalation:

Move exposed individual to fresh air. Loosen clothing as necessary and position individual in a comfortable position. Seek medical advice if discomfort or irritation persists. If breathing difficult, give oxygen.

#### **After skin contact:**

Wash affected area with soap and water. Rinse/flush exposed skin gently using water for 15-20 minutes. Seek medical advice if discomfort or irritation persists.

## After eye contact:

Protect unexposed eye. Rinse/flush exposed eye(s) gently using water for 15-20 minutes. Remove contact lens(es) if able to do so during rinsing. Seek medical attention if irritation persists or if concerned.

#### After swallowing:

Rinse mouth thoroughly. Do not induce vomiting. Have exposed individual drink sips of water. Seek medical attention if irritation, discomfort or vomiting persists.

## Most important symptoms and effects, both acute and delayed:

Irritation. Nausea. Headache. Shortness of breath.

## Indication of any immediate medical attention and special treatment needed:

If seeking medical attention, provide SDS document to physician.

## **SECTION 5: Firefighting measures**

#### **Extinguishing media**

according to 29CFR1910/1200 and GHS Rev. 3

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# **Cupric Chloride, Dihydrate**

#### Suitable extinguishing agents:

If in laboratory setting, follow laboratory fire suppression procedures. Use appropriate fire suppression agents for adjacent combustible materials or sources of ignition.

#### Unsuitable extinguishing agents: None

# Special hazards arising from the substance or mixture:

Combustion products may include carbon oxides or other toxic vapors. Avoid generating dust; fine dust dispersed in air in sufficient concentrations, and in the presence of an ignition source is a potential dust explosion hazard.

## Advice for firefighters:

## **Protective equipment:**

Use NIOSH-approved respiratory protection/breathing apparatus.

# Additional information (precautions):

Move product containers away from fire or keep cool with water spray as a protective measure, where feasible. Use spark-proof tools and explosion-proof equipment.

# **SECTION 6: Accidental release measures**

# Personal precautions, protective equipment and emergency procedures:

Wear protective equipment. Transfer to a disposal or recovery container. Use spark-proof tools and explosion-proof equipment. Use respiratory protective device against the effects of fumes/dust/aerosol. Keep unprotected persons away. Ensure adequate ventilation. Keep away from ignition sources.

## **Environmental precautions:**

Prevent from reaching drains, sewer or waterway. Collect contaminated soil for characterization per Section 13.

#### Methods and material for containment and cleaning up:

If in a laboratory setting, follow Chemical Hygiene Plan procedures. Place into properly labeled containers for recovery or disposal. If necessary, use trained response staff/contractor. Dust deposits should not be allowed to accumulate on surfaces, as these may form an explosive mixture if they are released into the atmosphere in sufficient concentration. Avoid dispersal of dust in the air. Collect solids in powder form using vacuum with HEPA filter.

# Reference to other sections: None SECTION 7: Handling and storage

#### Precautions for safe handling:

Minimize dust generation and accumulation. Wash hands after handling. Routine housekeeping should be instituted to ensure that dusts do not accumulate on surfaces. Dry powders can build static electricity charges when subjected to the friction of transfer and mixing operations. Follow good hygiene procedures when handling chemical materials. Do not eat, drink, smoke, or use personal products when handling chemical substances. If in a laboratory setting, follow Chemical Hygiene Plan. Use only in well ventilated areas. Avoid contact with skin, eyes and clothing.

## Conditions for safe storage, including any incompatibilities:

Provide ventilation for containers. Avoid storage near extreme heat, ignition sources or open flame. Store away from foodstuffs. Store away from oxidizing agents. Store in cool, dry conditions in well sealed containers. Store with like hazards.

## **SECTION 8: Exposure controls/personal protection**







according to 29CFR1910/1200 and GHS Rev. 3

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# **Cupric Chloride, Dihydrate**

Control parameters: , , OSHA PEL TWA (Total Dust) 15 mg/m³ (50 mppcf\*).

, , ACGIH TLV TWA (inhalable particles) 10 mg/m3.

10125-13-0, Cupric Chloride, Dihydrate, NIOSH TWA 1mg/m3.

Appropriate engineering controls:

Emergency eye wash fountains and safety showers should be available in the immediate vicinity of use/handling. Provide exhaust ventilation or other engineering controls to keep the airborne concentrations of vapor or dusts (total/respirable) below the applicable workplace exposure limits (Occupational Exposure Limits-OELs) indicated above. Use under a fume hood. It is recommended that all dust control equipment such as local exhaust ventilation and material transport systems involved in handling of this product contain explosion relief vents or an explosion suppression system or an oxygen deficient environment. Ensure that dust-handling systems (such as exhaust ducts, dust collectors, vessels, and processing equipment) are designed in a manner to prevent the escape of dust into

the work area (i.e., there is no leakage from the equipment).

**Respiratory protection:** Not required under normal conditions of use. Use suitable respiratory

protective device when high concentrations are present. For spills,

respiratory protection may be advisable.

**Protection of skin:** The glove material has to be impermeable and resistant to the product/

the substance/ the preparation being used/handled. Selection of the glove material on consideration of the penetration times, rates of diffusion and

the degradation.

**Eye protection:** Safety glasses with side shields or goggles.

**General hygienic measures:** The usual precautionary measures are to be adhered to when handling

chemicals. Keep away from food, beverages and feed sources.

Immediately remove all soiled and contaminated clothing. Wash hands

before breaks and at the end of work. Do not inhale

gases/fumes/dust/mist/vapor/aerosols. Avoid contact with the eyes and

skin.

## **SECTION 9: Physical and chemical properties**

Appearance (physical state, color):	Blue - green solid	Explosion limit lower: Explosion limit upper:	Not determined Not determined
Odor:	Odorless	Vapor pressure at 20°C:	Not determined
Odor threshold:	Not determined	Vapor density:	Not determined
pH-value:	3 - 3.8 (5% aq.sol.)	Relative density:	2.51 g/mL @ 20°
Melting/Freezing point:	100°C	Solubilities:	Material is water soluble.
Boiling point/Boiling range:	993°C	Partition coefficient (noctanol/water):	Not determined
Flash point (closed cup):	Not determined	Auto/Self-ignition temperature:	Not determined
Evaporation rate:	Not determined	Decomposition temperature:	Not determined
Flammability (solid, gaseous):	Not determined	Viscosity:	a. Kinematic: Not determined b. Dynamic: Not determined
Density at 20°C:	Not determined		
Cupric Chloride Dihydrate	Molecular Weight: 170.4788		

#### **SECTION 10: Stability and reactivity**

according to 29CFR1910/1200 and GHS Rev. 3

**Initial preparation date: : 01.21.2015** 

#### **Cupric Chloride, Dihydrate**

Reactivity: None Chemical stability:

No decomposition if used and stored according to specifications.

#### Possible hazardous reactions:

None under normal processing.

#### Conditions to avoid:

Store away from oxidizing agents, strong acids or bases. High temperatures, dust generation, exposure to moist air or water.

#### **Incompatible materials:**

Aqueous solution of copper (2+) sulfate is an acid. Incompatible with strong bases, hydroxylamine, magnesium.

#### **Hazardous decomposition products:**

Carbon oxides (CO, CO2). Oxides of sulfur, copper fumes.

## **SECTION 11: Toxicological information**

**Acute Toxicity**: None

**Chronic Toxicity**: No additional information.

Skin corrosion/irritation: No additional information.

Serious eye damage/irritation: No additional information.

Respiratory or skin sensitization: No additional information.

**Carcinogenicity**: No additional information.

**Germ cell mutagenicity**: No additional information. **Reproductive Toxicity**: No additional information.

STOT-single and repeated exposure: No additional information.

Additional toxicological information:

No additional information.

## **SECTION 12: Ecological information**

**Ecotoxicity:** No additional information. **Persistence and degradability**:

Readily degradable in the environment.

**Bioaccumulative potential**: No additional information.

**Mobility in soil**: No additional information.

Other adverse effects: No additional information.

#### **SECTION 13: Disposal considerations**

#### Waste disposal recommendations:

Product/containers must not be disposed together with household garbage. Do not allow product to reach sewage system or open water. It is the responsibility of the waste generator to properly characterize all waste materials according to applicable regulatory entities (US 40CFR262.11). Consult federal state/ provincial and local regulations regarding the proper disposal of waste material that may incorporate some amount of this product.

#### **SECTION 14: Transport information**

**US DOT** 

according to 29CFR1910/1200 and GHS Rev. 3

**Initial preparation date: : 01.21.2015** 

# **Cupric Chloride, Dihydrate**

UN Number:

ADR, ADN, DOT, IMDG, IATA 2802

Limited Quantity Exception: None

Bulk: Non Bulk:

RQ (if applicable): None RQ (if applicable): None

**Proper shipping Name:** COPPER CHLORIDE. **Proper shipping Name:** COPPER CHLORIDE.

Hazard Class: 8
Packing Group: |||.
Packing Group: |||.

Marine Pollutant (if applicable): No Marine Pollutant (if applicable): No

additional information. additional information.

Comments: None Comments: None





#### **SECTION 15: Regulatory information**

#### United States (USA)

# SARA Section 311/312 (Specific toxic chemical listings):

Acute

## SARA Section 313 (Specific toxic chemical listings):

10125-13-0 Cupric Chloride, Dihydrate.

#### RCRA (hazardous waste code):

None of the ingredients are listed.

#### TSCA (Toxic Substances Control Act):

All ingredients are listed.

## CERCLA (Comprehensive Environmental Response, Compensation, and Liability Act):

7447-39-4 Cupric chloride (anhydrous) 10 lbs.

10125-13-0 Copper and Compounds. For metals listed under CERCLA (copper), no reporting of releases of the solid form is required (dia >100micrometers); RQ applies to smaller particles No RQ assigned.

# Proposition 65 (California):

#### Chemicals known to cause cancer:

None of the ingredients are listed.

# Chemicals known to cause reproductive toxicity for females:

None of the ingredients are listed.

# Chemicals known to cause reproductive toxicity for males:

None of the ingredients are listed.

#### Chemicals known to cause developmental toxicity:

None of the ingredients are listed.

#### Canada

# Canadian Domestic Substances List (DSL) :

All ingredients are listed.

according to 29CFR1910/1200 and GHS Rev. 3

**Initial preparation date: : 01.21.2015** 

#### **Cupric Chloride, Dihydrate**

#### **SECTION 16: Other information**

This product has been classified in accordance with hazard criteria of the Controlled Products Regulations and the SDS contains all the information required by the Controlled Products Regulations. Note. The responsibility to provide a safe workplace remains with the user. The user should consider the health hazards and safety information contained herein as a guide and should take those precautions required in an individual operation to instruct employees and develop work practice procedures for a safe work environment. The information contained herein is, to the best of our knowledge and belief, accurate. However, since the conditions of handling and use are beyond our control, we make no guarantee of results, and assume no liability for damages incurred by the use of this material. It is the responsibility of the user to comply with all applicable laws and regulations applicable to this material.

**NFPA**: 2-0-1 **HMIS**: 2-0-1

GHS Full Text Phrases: None

## **Abbreviations and Acronyms:**

IMDG International Maritime Code for Dangerous Goods.

PNEC. Predicted No-Effect Concentration (REACH).

CFR Code of Federal Regulations (USA)

SARA Superfund Amendments and Reauthorization Act (USA).

RCRA. Resource Conservation and Recovery Act (USA).

TSCA. Toxic Substances Control Act (USA).

NPRI National Pollutant Release Inventory (Canada).

DOT US Department of Transportation.

IATA International Air Transport Association.

GHS Globally Harmonized System of Classification and Labelling of Chemicals.

ACGIH American Conference of Governmental Industrial Hygienists

CAS Chemical Abstracts Service (division of the American Chemical Society).

NFPA National Fire Protection Association (USA).

HMIS Hazardous Materials Identification System (USA).

WHMIS Workplace Hazardous Materials Information System (Canada).

DNEL Derived No-Effect Level (REACH).