according to 29CFR1910/1200 and GHS Rev. 3

Initial preparation date: : 01.08.2015

### Ferrous Ammonium Sulfate, 0.1M

## SECTION 1: Identification of the substance/mixture and of the supplier

**Product name**: Ferrous Ammonium Sulfate, 0.1M

Manufacturer/Supplier Article number: S25854

Recommended uses of the product and restrictions on use: Laboratory Chemicals

#### **Manufacturer Details:**

AquaPhoenix Scientific 860 Gitts Run Road, Hanover, PA 17331 (717) 632-1291

### **Supplier Details:**

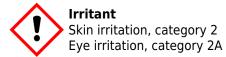
Fisher Science Education 6771 Silver Crest Road, Nazareth, PA 18064 800 955-1177

#### **Emergency telephone number:**

Emergency Telephone No.: 800-255-3924

### **SECTION 2: Hazards identification**

#### Classification of the substance or mixture:



Skin corrosion/irritation - Category 2. Eye Irritation 2.

Signal word: Warning

#### **Hazard statements:**

Causes skin irritation.

Causes serious eye irritation.

#### **Precautionary statements:**

If medical advice is needed have product container or label at hand.

Keep out of reach of children.

Read label before use.

Wear protective gloves/protective clothing/eye protection/face protection.

Wash skin thoroughly after handling.

IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses if present and easy to do. Continue rinsing.

If skin irritation occurs: Get medical advice/attention.

Take off contaminated clothing and wash before reuse.

Specific treatment (see supplemental first aid instructions on this label).

If eye irritation persists get medical advice/attention.

Store in a well ventilated place. Keep container tightly closed.

Store locked up.

#### Other Non-GHS Classification: None

according to 29CFR1910/1200 and GHS Rev. 3

**Initial preparation date: : 01.08.2015** 

#### Ferrous Ammonium Sulfate, 0.1M

### **SECTION 3: Composition/information on ingredients**

#### Ingredients:

Ingredients:				
CAS 7664-93-9	Sulfuric Acid, ACS	0.18 %		
CAS 7732-18-5	Deionized Water	95.9 %		
CAS 7783-85-9	Ferrous Ammonium Sulfate, ACS	3.92 %		
Percentages are by weight				

#### **SECTION 4: First aid measures**

#### **Description of first aid measures**

#### After inhalation:

Move exposed individual to fresh air. Loosen clothing as necessary and position individual in a comfortable position. Seek medical attention if irritation or coughing persists.

#### After skin contact:

Wash affected area with soap and water. Immediately remove contaminated clothing and shoes. Rinse thoroughly with plenty of water for at least 15 minutes. Immediately seek medical attention.

## After eye contact:

Protect unexposed eye. Flush thoroughly with plenty of water for at least 15 minutes. Remove contact lenses, if present and easy to do, and continue rinsing. Continue rinsing eyes during transport to hospital.

#### After swallowing:

Rinse mouth thoroughly. Dilute with water or milk. Get medical assistance.

## Most important symptoms and effects, both acute and delayed:

Inhalation may cause irritation to nose and upper respiratory tract, ulceration, coughing, chest tightness and shortness of breath. Higher concentrations cause tachypnoea, pulmonary oedema and suffocation. Pain, eye ulceration, conjunctival irritation, cataracts and glaucoma may occur following eye exposure. None identified.

### Indication of any immediate medical attention and special treatment needed:

Provide SDS to Physician. Physician should treat symptomatically.

#### **SECTION 5: Firefighting measures**

### **Extinguishing media**

## Suitable extinguishing agents:

Use dry chemical, foam, carbon dioxide, or mist to extinguish surrounding fire.

#### Unsuitable extinguishing agents:

None identified.

#### Special hazards arising from the substance or mixture:

May be flammable when in contact with combustible or strong reducing agents. Can react with metals to release flammable hydrogen gas. May react explosively with combustible organic or readily oxidizable materials.

#### Advice for firefighters:

#### **Protective equipment:**

Use normal procedures. Use protective clothing. Use NIOSHapproved breathing equipment.

# Additional information (precautions):

according to 29CFR1910/1200 and GHS Rev. 3

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Avoid inhaling gases, fumes, dust, mist, vapor, and aerosols. Avoid contact with skin, eyes, and clothing.

#### **SECTION 6: Accidental release measures**

#### Personal precautions, protective equipment and emergency procedures:

Ensure adequate ventilation. Ensure that air-handling systems are operational.

## **Environmental precautions:**

Should not be released into environment. Prevent from reaching drains, sewer, or waterway.

#### Methods and material for containment and cleaning up:

Always obey local regulations. If necessary use trained response staff or contractor. Evacuate personnel to safe areas. Containerize for disposal. Refer to Section 13. Keep in suitable closed containers for disposal. Soak up with inert absorbent material and dispose of as hazardous waste. Cover spill with suitable absorbing agent. Mix and add water to form slurry. Wear protective eyeware, gloves, and clothing. Refer to Section 8.

## Reference to other sections: None

## **SECTION 7: Handling and storage**

#### Precautions for safe handling:

Prevent formation of aerosols. Follow good hygiene procedures when handling chemical materials. Refer to Section 8. Prevent contact with skin, eyes, and clothing. Do not eat, drink, smoke, or use personal products when handling chemical substances.

# Conditions for safe storage, including any incompatibilities:

Store in a cool location. Keep away from food and beverages. Protect from freezing and physical damage. Keep out of direct sunlight and away from heat, water, and incompatible materials. Provide ventilation for containers. Keep container tightly closed.

# **SECTION 8: Exposure controls/personal protection**





**Control parameters:** 7783-85-9, Ferrous Ammonium Sulfate, OSHA PEL TWA (Total Dust) 15

mg/m3.

7783-85-9, Ferrous Ammonium Sulfate, ACGIH TLV TWA (inhalable

particles) 10 mg/m3.

7664-93-9, Sulfuric Acid., OSHA PEL: 1 mg/m³. 7664-93-9, Sulfuric Acid., ACGIH TLV: 0.2 mg/m³.

**Appropriate engineering controls:** Provide exhaust ventilation or other engineering controls to keep the

airborne concentrations of vapor and mists below the applicable

workplace exposure limits (Occupational Exposure Limits-OELs) indicated above. Emergency eye wash fountains and safety showers should be

available in the immediate vicinity of handling.

**Respiratory protection:** Not required under normal conditions of use. Where risk assessment

shows air-purifying respirators are appropriate use a full-face particle respirator type N100 (US) or type P3 (EN 143) respirator cartridges as a backup to engineering controls. When necessary use NIOSH approved

breathing equipment.

according to 29CFR1910/1200 and GHS Rev. 3

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**Protection of skin:** Select glove material impermeable and resistant to the substance. Select

glove material based on rates of diffusion and degradation. Dispose of contaminated gloves after use in accordance with applicable laws and good laboratory practices. Use proper glove removal technique without touching outer surface. Avoid skin contact with used gloves. Wear

protective clothing.

**Eye protection:** Tightly fitting safety goggles.

**General hygienic measures:** Perform routine housekeeping. Wash hands before breaks and

immediately after handling the product. Avoid contact with skin, eyes,

and clothing. Before rewearing wash contaminated clothing.

## **SECTION 9: Physical and chemical properties**

Appearance (physical state, color):	Clear, colorless liquid	Explosion limit lower: Explosion limit upper:	No information available No information available
Odor:	Odorless	Vapor pressure at 20°C:	No information available
Odor threshold:	No information available	Vapor density:	No Determined.
pH-value:	Not determined	Relative density:	Not available
, J.	No information available		Miscible.
Boiling point/Boiling range:	No information available	Partition coefficient (noctanol/water):	Not determined
Flash point (closed cup):	INIAT ANNIICANIA	Auto/Self-ignition temperature:	Not determined
Evaporation rate:		Decomposition temperature:	Not determined
Flammability (solid, gaseous):	No information available	Viscosity:	a. Kinematic: Not determined b. Dynamic: Not determined
Density at 20°C:	Approx.1 g/cm³ (8.345 lbs./gal) at 20°C (68°F)		
Hydrochloric Acid	MW is36.46		

## **SECTION 10: Stability and reactivity**

### Reactivity:

None under normal processing. May be flammable when in contact with combustible or strong reducing agents. Can react with metals to release flammable hydrogen gas. May react explosively with combustible organic or readily oxidizable materials.

#### **Chemical stability:**

Stable under normal conditions of use and storage.

## **Possible hazardous reactions:**

None under normal processing.

#### Conditions to avoid:

Incompatible materials.

#### **Incompatible materials:**

Chlorates, carbides, organics, picrates, fulminates, alkalines, nitrates, metals, reducing agents, oxidizing agents, acetic acid.

## **Hazardous decomposition products:**

Thermal decomposition can produce toxic fumes and oxides of sulfur.

according to 29CFR1910/1200 and GHS Rev. 3

Initial preparation date: : 01.08.2015

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### **SECTION 11: Toxicological information**

**Acute Toxicity**: No additional information. **Chronic Toxicity**: No additional information.

**Skin corrosion/irritation**: No additional information. **Serious eye damage/irritation**: No additional information.

Respiratory or skin sensitization:

None identified.

Carcinogenicity: No additional information.

**Germ cell mutagenicity**: No additional information. **Reproductive Toxicity**: No additional information.

**STOT-single and repeated exposure**: No additional information.

Additional toxicological information:

No additional information.

#### **SECTION 12: Ecological information**

### **Ecotoxicity:**

Gambusia affinis (Mosquito fish), 7664-93-9 LC50 - 42 mg/l - 96 h. Daphnia magna (Water flea) , 7664-93-9; EC50 29 mg/l - 24 h.

### Persistence and degradability:

No Information Available.

#### **Bioaccumulative potential:**

No Information Available.

# Mobility in soil:

No Information Available.

#### Other adverse effects:

No Information Available.

# **SECTION 13: Disposal considerations**

## Waste disposal recommendations:

Cover spill with sodium carbonate or soda ash-slaked lime mixture. Add water to form slurry. Flush to sewer with large quantities of water. Treat solid residue as normal refuse, unless prohibited due to iron content in the substance. Chemical waste generators must determine whether a discarded chemical is classified as a hazardous waste. Chemical waste generators must also consult local, regional, and national hazardous waste regulations. Ensure complete and accurate classification.

## **SECTION 14: Transport information**

**US DOT** 

**UN Number:** 

ADR, ADN, DOT, IMDG, IATA Not regulated

Limited Quantity Exception: None

according to 29CFR1910/1200 and GHS Rev. 3

**Initial preparation date:** : 01.08.2015

#### Ferrous Ammonium Sulfate, 0.1M

Bulk: Non Bulk:

RQ (if applicable): None RQ (if applicable): None

**Proper shipping Name:** Not regulated. **Proper shipping Name:** Not regulated.

Hazard Class: None Hazard Class: None

Packing Group: Not regulated.

Marine Pollutant (if applicable): No

Marine Pollutant (if applicable): No

additional information. additional information.

Comments: None Comments: None

## **SECTION 15: Regulatory information**

## **United States (USA)**

# SARA Section 311/312 (Specific toxic chemical listings):

None of the ingredients are listed.

# SARA Section 313 (Specific toxic chemical listings):

7664-93-9 Sulfuric Acid.

## RCRA (hazardous waste code):

None of the ingredients are listed.

# TSCA (Toxic Substances Control Act):

All ingredients are listed.

### CERCLA (Comprehensive Environmental Response, Compensation, and Liability Act):

7664-93-9 Sulfuric Acid 1000 lbs.

7783-85-9 Ferrous Ammonium Sulfate 1000 lbs.

## Proposition 65 (California):

#### Chemicals known to cause cancer:

None of the ingredients are listed.

## Chemicals known to cause reproductive toxicity for females:

None of the ingredients are listed.

### Chemicals known to cause reproductive toxicity for males:

None of the ingredients are listed.

### Chemicals known to cause developmental toxicity:

None of the ingredients are listed.

#### Canada

## Canadian Domestic Substances List (DSL) :

All ingredients are listed.

### **SECTION 16: Other information**

This product has been classified in accordance with hazard criteria of the Controlled Products Regulations and the SDS contains all the information required by the Controlled Products Regulations. Note. The responsibility to provide a safe workplace remains with the user. The user should consider the health hazards and safety information contained herein as a guide and should take those precautions required in an individual operation to instruct employees and develop work practice procedures for a safe work environment. The information

according to 29CFR1910/1200 and GHS Rev. 3

Initial preparation date: : 01.08.2015

### Ferrous Ammonium Sulfate, 0.1M

contained herein is, to the best of our knowledge and belief, accurate. However, since the conditions of handling and use are beyond our control, we make no guarantee of results, and assume no liability for damages incurred by the use of this material. It is the responsibility of the user to comply with all applicable laws and regulations applicable to this material.

**NFPA**: 1-0-1 **HMIS**: 1-0-1

GHS Full Text Phrases: None

# **Abbreviations and Acronyms:**

IMDG International Maritime Code for Dangerous	s Goods.
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PNEC. Predicted No-Effect Concentration (REACH).

CFR Code of Federal Regulations (USA)

SARA Superfund Amendments and Reauthorization Act (USA).

RCRA. Resource Conservation and Recovery Act (USA).

TSCA. Toxic Substances Control Act (USA).

NPRI National Pollutant Release Inventory (Canada).

DOT US Department of Transportation.

IATA International Air Transport Association.

GHS Globally Harmonized System of Classification and Labelling of Chemicals.

ACGIH American Conference of Governmental Industrial Hygienists

CAS Chemical Abstracts Service (division of the American Chemical Society).

NFPA National Fire Protection Association (USA).

HMIS Hazardous Materials Identification System (USA).

WHMIS Workplace Hazardous Materials Information System (Canada).

DNEL Derived No-Effect Level (REACH).