according to 29CFR1910/1200 and GHS Rev. 3

Initial preparation date: : 12.17.2014

Cupric Nitrate, Reagent Grade

SECTION 1: Identification of the substance/mixture and of the supplier

Product name: Cupric Nitrate, Reagent Grade

Manufacturer/Supplier Article number: S25282A

Recommended uses of the product and restrictions on use: Laboratory chemicals

Manufacturer Details:

AquaPhoenix Scientific 860 Gitts Run Road, Hanover, PA 17331 (717) 632-1291

Supplier Details:

Fisher Science Education 6771 Silver Crest Road, Nazareth, PA 18064 800 955-1177

Emergency telephone number:

Emergency Telephone No.: 800-255-3924

SECTION 2: Hazards identification

Classification of the substance or mixture:



Oxidizing

Oxidizing solids, category 2



Corrosive

Serious eye damage, category 1



Irritant

Acute toxicity (oral, dermal, inhalation), category 4 Skin irritation, category 2



Environmentally Damaging

Acute hazards to the aquatic environment, category 1

Oxidizing solids (Category 2).
Acute toxicity , Oral (Category 4).
Skin irritation (Category 2).
Serious eye damage (Category 1).
Acute aquatic toxicity (Category 1).

Signal word: Danger

Hazard statements:

May intensify fire; oxidizer. Harmful if swallowed. Causes skin irritation. Causes serious eye damage. Very toxic to aquatic life.

according to 29CFR1910/1200 and GHS Rev. 3

Initial preparation date: : 12.17.2014

Cupric Nitrate, Reagent Grade

Precautionary statements:

Wash skin thoroughly after handling.

Keep away from heat/sparks/open flames/hot surfaces. No smoking.

Keep/Store away from clothing/combustible materials.

Take any precaution to avoid mixing with combustibles.

Avoid release to the environment.

Do not eat, drink or smoke when using this product.

Wear protective gloves/protective clothing/eye protection/face protection.

Keep only in original container.

IF SWALLOWED: Call a POISON CENTER or doctor/physician if you feel unwell.

In case of fire, use agents recommended in section 5 for extinction.

Specific treatment (see supplemental first aid instructions on this label).

Collect spillage.

IF SWALLOWED: Rinse mouth. Do NOT induce vomiting.

IF ON SKIN: Wash with soap and water.

IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses if present and easy to do.

Continue rinsing.

If skin irritation occurs: Get medical advice/attention.

Take off contaminated clothing and wash before reuse.

Immediately call a POISON CENTER or doctor/physician.

Rinse mouth.

Dispose of contents and container as instructed in Section 13

Other Non-GHS Classification: None

SECTION 3: Composition/information on ingredients

Ingredients:

Ingredients:		
CAS 10031-43-3	Cupric Nitrate	100 %
	().	Percentages are by weight

SECTION 4: First aid measures

Description of first aid measures

After inhalation:

Move exposed individual to fresh air. Loosen clothing as necessary and position individual in a comfortable position. Seek medical advice if discomfort or irritation persists. If breathing difficult, give oxygen.

After skin contact:

Wash affected area with soap and water. Rinse thoroughly. Seek medical attention if irritation, discomfort or vomiting persists.

After eye contact:

Protect unexposed eye. Rinse/flush exposed eye(s) gently using water for 15-20 minutes. Remove contact lens(es) if able to do so during rinsing. Seek medical attention if irritation persists or if concerned.

After swallowing:

Rinse mouth thoroughly. Do not induce vomiting. Have exposed individual drink sips of water. Seek medical attention if irritation, discomfort or vomiting persists.

Most important symptoms and effects, both acute and delayed:

according to 29CFR1910/1200 and GHS Rev. 3

Initial preparation date: : 12.17.2014

Cupric Nitrate, Reagent Grade

Irritation. Causes severe respiratory and digestive tract irritation. May cause severe eye and skin irritation with possible burns. May cause ulceration of the conjunctiva and cornea. May cause severe irritation and possible burns. May cause dermatitis and skin discoloration. May cause severe gastrointestinal tract irritation with nausea, vomiting and possible burns. May cause hemorrhaging of the digestive tract. May cause severe irritation of the upper respiratory tract with pain, burns, and inflammation. May cause methemoglobinemia, cyanosis (bluish discoloration of skin due to deficient oxygenation of the blood), convulsions, tachycardia, dyspnea (labored breathing), and death. May cause ulceration and perforation of the nasal septum if inhaled in excessive quantities. Chronic inhalation and ingestion may cause effects similar to those of acute inhalation and ingestion. May cause liver and kidney damage. Individuals with Wilson's disease are unable to metabolize copper. Thus, copper accumulates in various tissues and may result in liver, kidney, and brain damage.

Indication of any immediate medical attention and special treatment needed:

If seeking medical attention, provide SDS document to physician. The use of d-Penicillamine as a chelating agent should be determined by qualified medical personnel.

SECTION 5: Firefighting measures

Extinguishing media

Suitable extinguishing agents:

If in laboratory setting, follow laboratory fire suppression procedures. Use appropriate fire suppression agents for adjacent combustible materials or sources of ignition.

Unsuitable extinguishing agents: None

Special hazards arising from the substance or mixture:

Combustion products may include carbon oxides or other toxic vapors. Thermal decomposition can lead to release of irritating gases and vapors. Avoid generating dust; fine dust dispersed in air in sufficient concentrations, and in the presence of an ignition source is a potential dust explosion hazard.

Advice for firefighters:

Protective equipment:

Use NIOSH-approved respiratory protection/breathing apparatus.

Additional information (precautions):

Move product containers away from fire or keep cool with water spray as a protective measure, where feasible. Use spark-proof tools and explosion-proof equipment.

SECTION 6: Accidental release measures

Personal precautions, protective equipment and emergency procedures:

Wear protective equipment. Transfer to a disposal or recovery container. Ensure adequate ventilation. Stop the spill, if possible. Contain spilled material by diking or using inert absorbent.

Environmental precautions:

Prevent from reaching drains, sewer or waterway. Collect contaminated soil for characterization per Section 13.

Methods and material for containment and cleaning up:

If in a laboratory setting, follow Chemical Hygiene Plan procedures. Place under an inert atmosphere. Provide ventilation. Place into properly labeled containers for recovery or disposal. If necessary, use trained response staff/contractor. Dust deposits should not be allowed to accumulate on surfaces, as these may form an explosive mixture if they are released into the atmosphere in sufficient concentration. Avoid dispersal of dust in the air. Vacuum or sweep up material and place into a suitable disposal container. Clean up spills immediately, observing precautions in the Protective Equipment section. Avoid generating dusty conditions.

Reference to other sections: None SECTION 7: Handling and storage

Precautions for safe handling:

according to 29CFR1910/1200 and GHS Rev. 3

Initial preparation date: : 12.17.2014

Cupric Nitrate, Reagent Grade

Wash thoroughly after handling. Use with adequate ventilation. Minimize dust generation and accumulation. Contents may develop pressure upon prolonged storage. Keep away from heat, sparks and flame. Avoid contact with clothing and other combustible materials. Do not get on skin or in eyes. Do not ingest or inhale. Routine housekeeping should be instituted to ensure that dusts do not accumulate on surfaces. Do not eat, drink, smoke, or use personal products when handling chemical substances.

Conditions for safe storage, including any incompatibilities:

Do not store near combustible materials. Store in a tightly closed container. Store in a cool, dry, well-ventilated area away from incompatible substances. Store away from foodstuffs. Keep away from heat, sparks, and flame. Store locked up.

SECTION 8: Exposure controls/personal protection





Control parameters:

Appropriate engineering controls:

10031-43-3, Copper(II) nitrate trihydrate, 1.00 mg/m3 TWA, USA NIOSH. Emergency eye wash fountains and safety showers should be available in the immediate vicinity of use/handling. Provide exhaust ventilation or other engineering controls to keep the airborne concentrations of vapor or dusts (total/respirable) below the applicable workplace exposure limits (Occupational Exposure Limits-OELs) indicated above. Use under a fume hood. It is recommended that all dust control equipment such as local exhaust ventilation and material transport systems involved in handling of this product contain explosion relief vents or an explosion suppression system or an oxygen deficient environment. Ensure that dust-handling systems (such as exhaust ducts, dust collectors, vessels, and processing equipment) are designed in a manner to prevent the escape of dust into

Respiratory protection:

Not required under normal conditions of use. Use suitable respiratory protective device when high concentrations are present. Use suitable respiratory protective device when aerosol or mist is formed. For spills, respiratory protection may be advisable.

the work area (i.e., there is no leakage from the equipment).

Protection of skin:

The glove material has to be impermeable and resistant to the product/ the substance/ the preparation being used/handled. Selection of the glove material on consideration of the penetration times, rates of diffusion and the degradation.

Eye protection:

Safety glasses with side shields or goggles.

General hygienic measures:

The usual precautionary measures are to be adhered to when handling chemicals. Keep away from food, beverages and feed sources.

Immediately remove all soiled and contaminated clothing. Wash hands

before breaks and at the end of work. Do not inhale $% \left\{ \left(n_{i}^{2}\right) \right\} =\left\{ \left(n_{i$

gases/fumes/dust/mist/vapor/aerosols. Avoid contact with the eyes and

skin.

SECTION 9: Physical and chemical properties

Appearance (physical state, color):	IRIIIA COIIA		Not determined Not determined
Odor:	Odorless	Vapor pressure at 20°C:	Not determined
Odor threshold:	Not determined	Vapor density:	8.33

according to 29CFR1910/1200 and GHS Rev. 3

Initial preparation date: : 12.17.2014

Cupric Nitrate, Reagent Grade				
pH-value:	4.0 for 0.2M soln.	Relative density:	2.05	
Melting/Freezing point:	114.5°C (238.1°F)	Solubilities:	Soluble.	
Boiling point/Boiling range:	INIAT ABTERMINEA	Partition coefficient (noctanol/water):	Not determined	
Flash point (closed cup):	Not determined	Auto/Self-ignition temperature:	Not determined	
Evaporation rate:		Decomposition temperature:	Not determined	
Flammability (solid, gaseous):	Not determined	Viscosity:	a. Kinematic: Not determined b. Dynamic: Not determined	
Density at 20°C:	Not determined			
Specific Gravity:	2.05 (Water = 1)			

SECTION 10: Stability and reactivity

Reactivity:

No data available.

Chemical stability:

No decomposition if used and stored according to specifications.

Possible hazardous reactions: None

Conditions to avoid:

Incompatible materials, combustible materials, reducing agents, organic matter.

Incompatible materials:

Ignites paper spontaneously in the presence of moisture. Reducing agents, combustible materials, potassium ferrocyanide, ether, tin. Organic materials. Powdered metals.

Hazardous decomposition products:

Nitrogen oxides (NOx), Copper oxides. Irritating and toxic fumes and gases, copper fumes.

SECTION 11: Toxicological information

Acute Toxicity: No additional information. **Chronic Toxicity**: No additional information.

Skin corrosion/irritation: No additional information.

Serious eye damage/irritation: No additional information.

Respiratory or skin sensitization: No additional information.

Carcinogenicity: No additional information.

Germ cell mutagenicity: No additional information. **Reproductive Toxicity**: No additional information.

STOT-single and repeated exposure: No additional information.

Additional toxicological information:

No additional information.

SECTION 12: Ecological information

Ecotoxicity: No additional information. **Persistence and degradability**:

Readily degradable in the environment.

Bioaccumulative potential: No additional information.

according to 29CFR1910/1200 and GHS Rev. 3

Initial preparation date: : 12.17.2014

Cupric Nitrate, Reagent Grade

Mobility in soil: No additional information.

Other adverse effects: No additional information.

SECTION 13: Disposal considerations

Waste disposal recommendations:

Product/containers must not be disposed together with household garbage. Do not allow product to reach sewage system or open water. It is the responsibility of the waste generator to properly characterize all waste materials according to applicable regulatory entities (US 40CFR262.11). Consult federal state/ provincial and local regulations regarding the proper disposal of waste material that may incorporate some amount of this product.

SECTION 14: Transport information

US DOT

UN Number:

ADR, ADN, DOT, IMDG, IATA 1477

Limited Quantity Exception: None

Bulk: Non Bulk:

RQ (if applicable): None RQ (if applicable): None

Proper shipping Name: NITRATES, INORGANIC, N.O.S. (Cupric Nitrate). **Proper shipping Name:** NITRATES, INORGANIC, N.O.S. (Cupric Nitrate).

Hazard Class: 5
Packing Group: ||.
Packing Group: ||.

Marine Pollutant (if applicable): No Marine Pollutant (if applicable): No

additional information. additional information.

Comments: None Comments: None





SECTION 15: Regulatory information

United States (USA)

SARA Section 311/312 (Specific toxic chemical listings):

Acute, Chronic, Reactive

SARA Section 313 (Specific toxic chemical listings):

10031-43-3 Cupric Nitrate.

10031-43-3 Copper compounds (N100).

RCRA (hazardous waste code):

None of the ingredients are listed.

TSCA (Toxic Substances Control Act):

10031-43-3 Not listed: Cupric Nitrate: not listed.

CERCLA (Comprehensive Environmental Response, Compensation, and Liability Act):

according to 29CFR1910/1200 and GHS Rev. 3

Initial preparation date: : 12.17.2014

Cupric Nitrate, Reagent Grade

10031-43-3 Copper compounds No RQ assigned. 10031-43-3 Cupric nitrate (3251-23-8) 100 lbs.

Proposition 65 (California):

Chemicals known to cause cancer:

None of the ingredients are listed.

Chemicals known to cause reproductive toxicity for females:

None of the ingredients are listed.

Chemicals known to cause reproductive toxicity for males:

None of the ingredients are listed.

Chemicals known to cause developmental toxicity:

None of the ingredients are listed.

Canada

Canadian Domestic Substances List (DSL):

10031-43-3 Not listed: Cupric Nitrate: not listed.

SECTION 16: Other information

This product has been classified in accordance with hazard criteria of the Controlled Products Regulations and the SDS contains all the information required by the Controlled Products Regulations. Note. The responsibility to provide a safe workplace remains with the user. The user should consider the health hazards and safety information contained herein as a guide and should take those precautions required in an individual operation to instruct employees and develop work practice procedures for a safe work environment. The information contained herein is, to the best of our knowledge and belief, accurate. However, since the conditions of handling and use are beyond our control, we make no guarantee of results, and assume no liability for damages incurred by the use of this material. It is the responsibility of the user to comply with all applicable laws and regulations applicable to this material.

NFPA: 1-0-0 **HMIS**: 1-0-0

GHS Full Text Phrases: None

Abbreviations and Acronyms:

IMDG International Maritime Code for Dangerous Goods.

PNEC. Predicted No-Effect Concentration (REACH).

CFR Code of Federal Regulations (USA)

SARA Superfund Amendments and Reauthorization Act (USA).

RCRA. Resource Conservation and Recovery Act (USA).

TSCA. Toxic Substances Control Act (USA).

NPRI National Pollutant Release Inventory (Canada).

DOT US Department of Transportation.

IATA International Air Transport Association.

GHS Globally Harmonized System of Classification and Labelling of Chemicals.

ACGIH American Conference of Governmental Industrial Hygienists

CAS Chemical Abstracts Service (division of the American Chemical Society).

NFPA National Fire Protection Association (USA).

HMIS Hazardous Materials Identification System (USA).

according to 29CFR1910/1200 and GHS Rev. 3

Initial preparation date: : 12.17.2014

Cupric Nitrate, Reagent Grade

WHMIS Workplace Hazardous Materials Information System (Canada). SAMPLE: NOT FOR COMMERCIAL USE

DNEL Derived No-Effect Level (REACH).